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Types of Reaction Worksheet: Practice your six types of reaction. Types of Chemical Reaction II (dd-ch): More practice with figuring out the types of chemical reaction. Types of Chemical Reaction III (dd-ch): You can probably guess what this is on your own. Oxidation and Reduction Practice: In which we focus on oxidation states and redox reactions. This entry was posted in Uncategorized.

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Chemistry

Six Types of Chemical Reaction Worksheet

Answers Balance the following reactions and indicate which of the six types of chemical reaction are being represented: 1) $2 \text{NaBr} + 1 \text{Ca(OH)}_2 \rightarrow 1 \text{CaBr}_2 + 2 \text{NaOH}$ Type of reaction: double displacement 2) $2 \text{NH}_3 + 1 \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$ Type of reaction: synthesis 3) $4 \text{C}_5\text{H}_9\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$ Type of reaction: combustion

Six types of chemical reaction worksheet

Six Types of Reaction Practice Sheet Answers

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Using your knowledge of the six types of chemical reaction, balance the following equations and indicate which of the six types of reaction they are:

1) $2 \text{LiH} + 1 \text{F}_2 \rightarrow 2 \text{LiF} + 1 \text{H}_2$ Type of reaction: Single replacement

2) $2 \text{Fe}(\text{OH})_2 \rightarrow 1 \text{Fe}_2\text{O}_3 + 3 \text{H}_2\text{O}$ Type of reaction: Decomposition

3) $3 \text{H}_2\text{SO}_4 + 2 \text{Ga}(\text{OH})_3 \rightarrow 6 \text{H}_2\text{O} + 1 \text{Ga}_2(\text{SO}_4)_3$ Type of reaction: Acid-base

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Reserved Types of Reactions Worksheet -
Solutions 1) $3 \text{ NaBr} + 1 \text{ H}_3 \text{ PO}_4 + 1 \text{ Na}_3 \text{ PO}_4 + 3 \text{ HBr}$ Reaction type: _____ double displacement
2) $3 \text{ Ca(OH)}_2 + 1 \text{ Al}_2 (\text{SO}_4)_3 + 3 \text{ CaSO}_4 + 2 \text{ Al(OH)}_3$

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Combination reaction; Decomposition reaction; Displacement reaction; Double Displacement reaction; Precipitation

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Another very common chemical reaction is of two types, i.e. single displacement and double displacement. In single displacement reaction, any one chemical partner exchanges from reactants to products while two sets of chemical partners exchanges from reactants to products. An example for single displacement reaction is $XY + Z \rightarrow XZ + Y$.

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Updated 2-10-16. Types of reaction and reaction products: Overview of chemical reactions: Some info from a book I never finished about chemical reactions. Pretty much covers the stuff above. Types of reaction: There are several categories into which chemical reactions can be placed, and knowing them can help you to figure out how to predict the ways in which chemical reactions will operate.

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Types of Reactions Worksheet Balance the following equations and indicate the type of reaction taking place: H_3P_04 HBr Na_3P_04 + NaBr + $\text{Al}_2(\text{S}_04)_3$ Type of reaction: CaS_04 + Type of reaction: Mg + C_2H_4 + PbS_04 NH_3 + H_2O

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+ 02 PbS03 + Mgo Fe + Type of reaction: H2O
CO2 + Type of reaction: Type of reaction: H2
Type of reaction: 12 S03 N216 + H2S04

*LNaNO3+ PbS04 C3H8+ CO2 1-120 zn+ LHCI+ +1
10. 11. 12. 13 ...

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answers ... Predicting Reaction Products -
Solutions Balance the equations and predict
the products for the following reactions: 1)
 $3 \text{ Na} + 1 \text{ FeBr}_3 \rightarrow 3 \text{ NaBr} + 1 \text{ Fe}$ 2) $2 \text{ NaOH} + 1 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ Na}_2\text{SO}_4 + 2 \text{ H}_2\text{O}$

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